

Xanthophyll Esters in Fruits and Vegetables

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Abstract

Carotenoids possessing hydroxyl groups (xanthophylls) are often found as fatty acid esters in many fruits and vegetables. The developments in high resolution chromatographic and spectroscopic techniques have led to a detailed characterization of xanthophyll esters in commonly consumed fruits and vegetables, such as apples, apricots, mandarins, mangoes, papayas, red and chili peppers, potatoes or squash. Some more rich sources have been identified, like wolfberry (goji), sea buckthorn, persimmon, whose popularity is increasing due to the high content of bioactive compounds. Esterification increases the lipophilicity of xanthophylls and contributes to the sequestration of carotenoids, to the formation of specialized structures in the chromoplasts and to an increased photoprotection. The process occurs during ripening in fruits and it is associated with a significant change in colour. Even if the specific enzymes which catalyze the esterification process were not characterized yet in fruits, detailed analytical data regarding the carotenoid composition suggested a selectivity of these enzymes for certain fatty acids and selectivity for the ring in the case of non-symmetric xanthophylls. Xanthophyll esters seem to be efficiently hydrolyzed and absorbed in humans leading to a comparable bioavailability to the unesterified compounds. In addition, the xanthophyll esters preserve the antioxidant capacity of the parent compounds while having a better stability in fruits during storage and processing. All these properties are important from the perspective of the use of fruits rich in xanthophyll esters as valuable components of the human diet and as sources of bioactive compounds in the prevention of severe degenerative diseases.

Keywords: analysis, antioxidant capacity, biosynthesis and accumulation, bioavailability, stability

Introduction

Fruits and vegetables are important components of the human diet and they are associated with numerous health benefits due to their high content of micronutrients. Carotenoids are a family of widely distributed lipophilic compounds, comprising more than 750 representatives (Britton *et al.*, 2004). The yellow, orange or red colour of fruits and vegetables is due to the presence of carotenoid pigments. Carotenoids are biosynthesized by plants and some microorganisms but not by animals and human, which rely on their diet to incorporate carotenoids. It is considered that up to 90% of the carotenoid intake in humans comes from fruits and vegetables (Maiani *et al.*, 2009).

The two classes of carotenoids are carotenes and xanthophylls, which are oxygenated derivatives. Of particular importance among xanthophylls are the hydroxycarotenoids: β -cryptoxanthin, lutein and zeaxanthin. Some other well known xanthophylls are violaxanthin and neoxanthin in green tissues, capsanthin and capsorubin in pepper, astaxanthin in algae (*Haematococcus pluvialis*) or pink-flesh fish (Britton and Khachik, 2009; Dufossé, 2009). In fruits and vegetables xanthophylls are present either in a free, unesterified form, or as esters with fatty acids. In some foods like corn, spinach,

broccoli or other green leafy vegetables, xanthophylls are present exclusively in unesterified form. In other fruits and vegetables such as pepper, wolfberry, sea buckthorn, apple, squash etc., xanthophylls are mostly found in esterified form (Britton and Khachik, 2009; Pérez-Gálvez and Minguez Mosquera, 2005).

Analysis of carotenoids is usually performed on saponified extracts. Saponification (alkaline hydrolysis) is used to facilitate carotenoid isolation because it is effective in removing contaminating lipids (especially in food-rich samples) and destroying chlorophylls. In the same time, the hydrolysis of carotenoid esters occurs and artefacts can be produced. Due to the fact that xanthophylls present in fruits are often esterified, more and more studies are investigating the unsaponified carotenoid extracts which reflect the native composition of the plant food (Giuffrida *et al.*, 2013; Pinteá, 2008; Rodríguez-Amaya, 2010).

The best documented function of carotenoids is the provitamin A activity. Carotenoids possessing at least one unsubstituted β -ring are absorbed and cleaved to form retinol in animals and humans. Food rich in carotenoids are considered to be beneficial in the prevention of severe diseases like cancer, cardiovascular diseases, degenerative diseases, age-related macular degeneration (AMD) and cataract (Krinsky and Johnson, 2005; Landrum *et al.*, 1997; Rao and Rao, 2007). Lutein and zeaxanthin

accumulate in the structures of the human retina, with the highest concentration in the macula where they act as light filter pigments but also as antioxidants (Bernstein *et al.*, 2001). Observational, epidemiological and intervention trials demonstrated that a high concentration of lutein in plasma is correlated with an increase of macular pigment density and a reduced risk of AMD (Chew *et al.*, 2013; Schalch *et al.*, 2009). β -Cryptoxanthin has provitamin A activity but has also been proved to stimulate bone calcification (Yamaguchi, 2012), to have antiproliferative (Wu *et al.*, 2013) and anti-inflammatory effects (Pattison *et al.*, 2005).

Taking into account that xanthophylls esters are efficiently absorbed and cleaved into human body and that their availability is comparable with the unesterified form, food rich in esters can be considered a valuable source of biologically active xanthophylls. Furthermore, the xanthophyll esters preserve the well known antioxidant capacity of carotenoids which is associated with most of their biological properties.

Occurrence

Carotenoid esters occur in flowers, fruits and vegetables. In flower petals, xanthophylls are usually present in esterified form, the best known example being the flowers of *Tagetes* sp., where more than 90% of total carotenoids are lutein esters (Gregory *et al.*, 1986; Ohmiya, 2013; Pérez-Gálvez and Minguéz-Mosquera, 2005). Lutein, β -cryptoxanthin, zeaxanthin and violaxanthin are the most frequent xanthophylls found in esterified form in fruits and vegetables and they are located in the specialized structures in chromoplasts (Breithaupt and Bamedi, 2001; Ohmiya, 2013).

The hydroxyl groups of xanthophylls are acylated with saturated fatty acids such as lauric (C12:0), myristic (C14:0), palmitic (C16:0) and stearic (C18:0) acids (Britton *et al.*, 1995a) but unsaturated fatty acids like oleic acid (C18:1), linoleic (C18:2) or linolenic (C18:3) acids were also occasionally reported in fruits, as well as the short chain fatty acids like butyric acid (C4:0).

The most common fruits and vegetables which contain remarkable amounts of xanthophyll esters are: apples (*Malus domestica*) (Delgado-Pelayo *et al.*, 2014); apricots (*Prunus armeniaca* L.) (Kurz *et al.*, 2008); mangoes (*Mangifera indica* L.) (Ornelas-Paz *et al.*, 2007), sea

buckthorn berries (*Hippophae rhamnoides* L.) (Giuffrida *et al.*, 2012; Pop *et al.*, 2014; Weller and Breithaupt, 2003); mandarins (*Citrus reticulata*) (Giuffrida, 2006); gojis (*Lycium barbarum* L. and *Lycium chinense* L.) (Inbaraj *et al.*, 2008; Zhao *et al.*, 2013); papayas (*Carica papaya* L.) (Schweiggert *et al.*, 2012); oranges (*Citrus sinensis* L.) (Giuffrida *et al.*, 2010); orange, red and chilli peppers (*Capsicum* sp.) (Giuffrida *et al.*, 2013; Hornero-Méndez and Minguéz-Mosquera, 2000); potatoes (*Solanum* sp.) (Breithaupt and Bamedi, 2002a; Burmeister *et al.*, 2013; Fernandez-Orozco *et al.*, 2013).

Some other less common fruits which were also reported to contain xanthophyll esters are Chinese lanterns (*Physalis alkekengi* L.) (Goodwin, 1980; Pintea *et al.*, 2005; Weller and Breithaupt, 2003); persimmon (*Diospyros kaki*) (Weller and Breithaupt, 2003); yellow raspberries (*Rubus idaeus* L.) (Carvalho *et al.*, 2013); sarsaparilla berries (*Smilax aspera* L.) (Delgado-Pelayo and Hornero-Méndez, 2012); corozo, sastra, frutita, maracuja chino, mamey roja (Murillo *et al.*, 2013); cashew apple (Pinto de Abreu *et al.*, 2013).

A very complex screening regarding β -cryptoxanthin esters was performed by Breithaupt and Bamedi (2001). β -cryptoxanthin laurate, myristate and palmitate, as well as the free β -cryptoxanthin, were quantified in several fruits: orange, blood orange, pepper and chili, clementine, papaya, persimmon, tangerine, loquat, kumkuat, nectarine. Similarly, Weller and Breithaupt (2003) determined the zeaxanthin esters (mono and diesters with laurate, myristate, palmitate and stearate) in red and orange pepper, sea buckthorn, wolfberries, Chinese lanterns and persimmon. In both studies, LC/MS/APCI was used for separation and identification, as well as standard esters obtained by semisynthesis.

Carotenoid composition and concentration in fruits is influenced by genetic and environmental factors, such as: cultivar, maturity, climate, geographic area, processing and storage (Rodríguez-Amaya, 2010). Absolute quantitative data for the carotenoid esters content are difficult to be set. However, pepper and chilli (orange, red), wolfberry, papaya, tangerines, persimmon and sea buckthorn seem to be among the richest sources (Breithaupt and Bamedi, 2001; Weller and Breithaupt, 2003). Detailed composition of xanthophyll esters in selected fruits, including the fatty acids esterifying the hydroxyl groups are presented in Tab. 1.

Tab. 1. Distribution, extraction and analytical procedures for separation and identification of xanthophyll esters in selected fruits

Fruit	Analyte	Extraction protocol	Column (stationary phase)	Mobile phase	Detection system	References
Capsicum varieties	<i>Cis</i> -capsanthin-esters: C12:0; C14:0 Antheraxanthin esters: C12:0; C14:0 Capsanthin-5,6-epoxy esters: C14:0 Lutein esters: C14:0	Acetone	YMC C30 (250x4.6 mm, 5 μ m)	(A): MeOH:TBME:H ₂ O 82:16:2 (v/v/v)	HPLC- DAD- APCI-MS	Giuffrida <i>et al.</i> , 2013
	Capsanthin esters: C12:0; C14:0; C16:0; C14:0-C14:0; C12:0-C16:0; C14:0-C16:0; C16:0-C16:0 Zeaxanthin esters: C12:0; C14:0; C16:0; C12:0-C12:0; C12:0-C14:0; C14:0-C14:0; C14:0-C16:0; C16:0-C16:0 β -cryptoxanthin esters: C12:0; C14:0; C16:0 Cytocapsin esters: C14:0; C16:0 Capsorubin diesters: C14:0-C14:0; C14:0-C16:0			(B): MeOH:TBME:H ₂ O 10:88:2 (v/v/v)		

Spice paprika (<i>Capsicum annuum</i> L.)	Capsorubin methyl ester Cucurbitaxanthin methyl ester Capsanthin epoxide methyl ester Capsanthin methyl ester-1 Cryptoxanthin methyl ester Capsanthin ME-2ME Antheraxanthin ME Zeaxanthin ME and DE Capsanthin DE	Dichlorethane Acetone/ Methanol (2:1:1, v/v/v)	Purospher C-18 (250x4.6 mm, 3 µm)	A: H ₂ O B: MeOH C: Methanol/Isopropanol / Acetonitrile 10:55:35 (v/v/v)	LC-MS	Daoud, et al., 2014
Mango	9-cis-violaxanthin esters: C4:0; C4:0-C4:0 all-trans-violaxanthin: C4:0-C4:0; C4:0-C6:0	Methanol	YMC C30 (150 x 4.6 mm, 3 µm) 15°C	A: Water B: MeOH C: MTBE	LC-APCI+ -MS	Ornelas-Paz et al., 2007
Sea buckthorn (<i>Hippophae rhamnoides</i> L.)	Lutein esters: C16:3; C16:2; C18:2; 18:1; C16:0-C16:0 Zeaxanthin esters: C14:0; C16:0; C14:0-14:0; C16:0-16:1; C14:0-16:0; C16:0-16:0 β-cryptoxanthin esters: C14:0; C16:0	Methanol/ Ethyl Acetate/ Petroleum Ether, (1/1/1; v/v/v)	YMC C30 (250 x 4.6 mm -5 µm)	A: Methanol/MTBE/H ₂ O (83:15:2, v/v/v) B: Methanol/MTBE/H ₂ O (8:90:2, v/v/v)	HPLC- DAD- APCI-MS	Giuffrida et al, 2012
	β-cryptoxanthin esters: C16:0 Zeaxanthin esters: C14:0; C16:0; C14:0-C16:0; C16:0-C16:0 Lutein esters: C16:0; C14:0-14:0; C14:0-16:0; C16:0-C18:0	Methanol/ Ethyl Acetate/ Petroleum Ether, (1/1/1; v/v/v)	LiChrosorb RP 18 (250 x 4.6 mm, 5 µm) RP 18 UPLC BEH Shield (2.1 x 150 mm, 1.7 µm)	A: AcCN:H ₂ O 9:1 B: AcEt A: Ac Ethyl B: Acetonitrile C: Acetonitrile: H ₂ O (1:1, v/v)	RP-PDA- HPLC UHPLC- PAD-ESI- MS	Pop et al., 2014
Mandarin essential oil	β-cryptoxanthin esters: C12:0; C14:0; C16:0	Oil:TBME (1:1)	YMC C30 (250x4.6 mm, 5 µm)	A: MeOH B: MTBE	HPLC- PDA-APCI- MS	Giuffrida et al., 2006
Sarsaparilla berries (<i>Smilax aspera</i> L.)	Zeaxanthin monoester: C14:0 β-cryptoxanthin esters: C10:0; C12:0; C14:0; C16:0; C18:0	acetone	RP C18 Mediterranea SEA 18 (200 x 4.6 mm, 3 µm); 25°C	A: Acetone B: H ₂ O	LC-MS (APCI+)	Delgado- Pelayo and Hornero- Méndez, 2012
Mandarin	β-cryptoxanthin esters: C10:0; C12:0; C14:0; C16:0 Mutatoxanthin isomer esters: C10:0; C12:0 Lutein esters: C14:0; C16:0; C18:0; C18:1 Mutatoxanthin esters: C16:0; C18:0 Luteoxanthin esters: C10:0; C12:0; C14:0	Essential oil injected directly	Discovery Cyano (250x1 mm; 5µm) Chromolith Performance RP- 18 (100x4.6 mm)	A: n-hexane B: n-hexane/ butylacetat/acetone (80:15:5) v/v/v A: 2-propanol B: 20% water in acetonitrile (v/v)	LC x LC, DAD/ APCI	Dugo et al., 2008
Orange juices	cis-violaxanthin esters: C10:0; C12:0; C14:0; C16:0; C12:0- C12:0; C12:0-C14:0; C14:0- C14:0; C14:0-C16:0; C16:0- C16:0 Luteoxanthin esters: C12:0; C14:0; C16:0 Auroxanthin: C16:0 Antheraxanthin: C16:0 Mutatoxanthin (C16:0) β-cryptoxanthin (C12:0; C14:0; C16:0)	Methanol/ Ethyl Acetate/ Petroleum Ether, (1/1/1; v/v/v)	YMC C30 (250x4.6 mm, 5 µm) 23°C	A: Methanol/MTBE/Wa ter (80:15:5) v/v/v B: Methanol/MTBE (10:90; v/v)	LC-MS- APCI	Giuffrida et al., 2010
Apricots (<i>Prunus armeniaca</i> L.)	β-cryptoxanthin esters (C12:0; C18:1; C16:0; C18:1-C18:1; C16:0-C18:1) Lutein esters (C18:1- C18:1; C16:0- C18:1; C12:0- C12:0; C12:0- C14:0; C14:0- C14:0; C14:0- C16:0; C16:0- C16:0) Antheraxanthin esters (C12:0; C14:0; C16:0)	Acetone/ Hexane (1:1; v/v)	C30 YMC 150x3 mm 3µm 25°C	A: Methanol/MTBE/ H ₂ O (81:15:4 v/v/v) B: Methanol/MTBE/H ₂ O (49:2:4 v/v/v)	LC-MS	Kurz et al., 2008
Yellow raspberries (<i>Rubus idaeus</i> L.)	Lutein mono and diesters: C8:0; C10:0; C12:0; C14:0; C16:0	Methanol/Tris- HCl buffer/ Chloroform (1:1:4; v/v/v)	C30 YMC (250 x 4.6 mm, 5 µm)	A: MeOH B: H ₂ O/MeOH (20/80 v/v) +0,2% ammonium acetate C: TMBE	LC-APCI- MS	Carvalho et al., 2013
Apple (flesh and peel)	all-trans-Neoxanthin esters: C10:0-C10:0; C10:0-C12:0; C12:0-C12:0; C12:0-C14:0; C14:0; C14:0-C14:0; C16:0; C16:0-C16:0; C16:0-C18:0; C16:0-C18:0 all-trans-Violaxanthin esters: C10:0-C10:0; C10:0-C12:0; C12:0-C12:0; C12:0-C14:0; C14:0-C14:0; C14:0-C16:0; C16:0-C16:0; C16:0-C18:0 9-cis-Violaxanthin esters: C10:0-C10:0; C10:0-C12:0; C12:0- C12:0; C12:0-C14:0; C14:0-C14:0; C14:0-C16:0; C16:0- C16:0; C16:0-C18:0	N,N-dimethyl- formamide	RP-C18 Mediterranea SEA 18 (200x4.6 mm, 3 µm) 25 °C	A: Acetone B: Deionised water	LC-MS (APCI+)	Delgado- Pelayo et al., 2014
Goji (<i>Lycium barbarum</i>)	Zeaxanthin esters: C16:0; C16:0-C16:0 β-cryptoxanthin esters: C16:0	Hexane/ Ethanol/ Acetone Toluene (10:6:7:7, v/v/v/v)	C30 25°C	A: Dichloromethane B: Methanol/Acetonitrile / Water (84:14:5; v/v/v)	HPLC- DAD- APCI-MS	Inbaraj et al., 2008

Corozo (<i>A. aculeate</i>)	Zeaxanthin esters (C10:0; C12:0; C16:0; C10:0-C10:0; C10:0-C12:0; C12:0-C12:0; C12:0-C14:0; C10:0-C16:0; C14:0-C14:0; C12:0-C16:0; C14:0-C16:0; C16:0-C16:0) β -criptoxanthin esters (C8:0; C10:0; C12:0; C14:0; C16:0)	Acetone	YMC C30 (250 x 4.6 mm, 5 μ m)	A: Methanol/MTBE/H ₂ O (81:17:2, v/v/v) B: Methanol/MTBE/H ₂ O (10:88:2, v/v/v)	HPLC-DAD-APCI-MS	Murillo <i>et al.</i> , 2013
Sastra (<i>G. intermedia</i>)	β -criptoxanthin esters (C12:0; C14:0; C16:0) Lutein esters (C12:0-C12:0; C12:0-C14:0; C12:0-C16:0) Zeaxanthin esters (C14:0-C14:0; C14:0-C16:0; C16:0-C16:0)	Acetone	YMC C30 (250 x 4.6 mm, 5 μ m)	A: Methanol/MTBE/H ₂ O (81:17:2, v/v/v) B: Methanol/MTBE/H ₂ O (10:88:2, v/v/v)	HPLC-DAD-APCI-MS	Murillo <i>et al.</i> , 2013
Sapote (<i>Q. cordata</i>)	n.i. (xanthophyll esters) Zeaxanthin esters (C12:0-C12:0; C12:0-C14:0; C14:0-C18:1; C14:0-C14:0; C12:0-C16:0; C16:0-C18:1; C14:0-C16:0; C16:0-C16:0; C16:0-C18:0; C18:0-C18:0)	Acetone	YMC C30 (250 x 4.6 mm, 5 μ m)	A: Methanol/MTBE/H ₂ O (81:17:2, v/v/v) B: Methanol/MTBE/H ₂ O (10:88:2, v/v/v)	HPLC-DAD-APCI-MS	Murillo <i>et al.</i> , 2013
Frutita <i>A. psilospermus</i>	n.i. (cis-Apo-carotenoid-ester) (C8:0; C12:0; C14:0; C16:0) β -citaurin (C6:0; C8:0; C10:0; C12:0; C14:0; C16:0)	Acetone	YMC C30 (250 x 4.6 mm, 5 μ m)	A: Methanol/MTBE/H ₂ O (81:17:2, v/v/v) B: Methanol/MTBE/H ₂ O (10:88:2, v/v/v)	HPLC-DAD-APCI-MS	Murillo <i>et al.</i> , 2013
Maracuja chino (<i>C. macrantis</i>)	n.i. (xanthophyll esters) Cryptocapsin (C12:0) β -criptoxanthin esters (C12:0; C14:0; C16:0)	Acetone	YMC C30 (250 x 4.6 mm, 5 μ m)	A: Methanol/MTBE/H ₂ O (81:17:2, v/v/v) B: Methanol/MTBE/H ₂ O (10:88:2, v/v/v)	HPLC-DAD-APCI-MS	Murillo <i>et al.</i> , 2013
Mamey rojo (<i>P. sapata</i>)	Cryptocapsin-5,6-epoxide ester: C12:0 β -criptoxanthin-5,8-epoxide ester: C12:0; 13z/13z-Cryptocapsin ester: C14:0 Cryptocapsin esters: (C14:0; C16:0; C18:0)	Acetone	YMC C30 (250 x 4.6 mm, 5 μ m)	A: Methanol/MTBE/H ₂ O (81:17:2, v/v/v) B: Methanol/MTBE/H ₂ O (10:88:2, v/v/v)	HPLC-DAD-APCI-MS	Murillo <i>et al.</i> , 2013
Papaya (<i>Carica papaya</i> L.)	β -criptoxanthin ester: C10:0; C12:0; C14:0; C16:0	Methanol/ Ethyl Acetate/ Petroleum Ether, (1/1/1; v/v/v) 25 °C	YMC C30 (150 x 3 mm, 3 μ m)	A: Methanol/MTBE/H ₂ O (91:5:4, v/v/v) B: Methanol/MTBE/H ₂ O (6:90:4, v/v/v)	HPLC-DAD-APCI	Schweigge rt <i>et al.</i> , 2012
Mamey (<i>Pouteria sapota</i>)	Lutein ester: C12:0-C14:0	Hexane/ Dichlorom etane (1:1, v/v)	YMC C30 (150 x 4.6 mm)	A: Methanol B: MTBE	HPLC LC-MS- TOF	Yahia <i>et al.</i> , 2011
Peach	β -criptoxanthin esters: C12:0; C14:0; C16:0	MeOH:Et Ac:PE (1:1:1, v/v/v)	YMC C30 (250 x 4.6 mm, 5 μ m)	A: Methanol/MTBE/H ₂ O (83:15:2, v/v/v) B: Methanol/MTBE/H ₂ O (8:90:2, v/v/v)	HPLC-DAD-APCI-MS	Giuffrida <i>et al.</i> , 2013
Tamarillo fruits (<i>Solanum betaceum Cav.</i>)	<i>All-trans</i> -neoxanthin esters: C14:0-C14:0; C14:0-C16:0; C16:0-C16:0 <i>All-trans</i> -5,6-epoxy- β -criptoxanthin esters: C14:0; C16:0 <i>Cis</i> -neoxanthin esters: C14:0-C14:0; C14:0-C16:0 <i>All-trans</i> -violaxanthin esters: C14:0-C14:0; C14:0-C16:0; C16:0-C16:0 <i>All-trans</i> - β -criptoxanthin esters: C14:0; C16:0 <i>All-trans</i> -antheraxanthin esters: C14:0-C14:0; C14:0-C16:0; C16:0-C16:0 <i>All-trans</i> -lutein esters: C14:0-C14:0; C14:0-C16:0 <i>All-trans</i> -zeaxanthin: C14:0-C14:0; C14:0-C16:0; C16:0-C16:0	Ethanol/H exane (4:3, v/v)	YMC C30 (250 x 4.6 mm, 5 μ m)	A: Water/20 mM ammonium acetate B: MeOH/20 mM ammonium acetate C: MTBE	HPLC/ PDA-MS	Mertz <i>et al.</i> , 2010

Structure and properties

Carotenoids are isoprenoid compounds which can be divided into carotenes (hydrocarbons) and oxygenated derivatives (xanthophylls). The xanthophylls of higher plants possess a C₄₀ backbone which can bear different oxygenated functional groups: hydroxyl, aldehyde, ketone, epoxy, carboxyl. Xanthophylls containing hydroxyl groups, also named carotenol, can be esterified with carboxylic acids, mainly with long chain and saturated ones. The chemical structure of the most common xanthophyll esters is presented in Fig. 1.

Most of the physical and chemical properties of carotenoids are due to the presence of the polyenic system. The long system of conjugated double bonds (chromophore) is responsible for the light absorption properties of carotenoids, in consequence also for their colouring properties. Also, the polyene chain is responsible for most of the chemical properties of carotenoids.

Esterification of hydroxyl group does not modify the chromophore, in consequence the UV-VIS light absorption spectra of the esters will be identical with those of the unesterified compounds (Britton, 1995; Britton *et al.*, 2004). However, the molar absorptivity (as well as the specific absorption coefficient $A^{1\%}_{1\text{cm}}$) is different. The molar absorptivity of xanthophyll ester can be calculated as the product between the specific absorption coefficient $A^{1\%}_{1\text{cm}}$ of the unesterified xanthophyll and the ratio between the molecular weight of xanthophyll and the molecular weight of the ester (Schiedt and Liaaen-Jensen, 1995). The similar light absorption properties of esterified and free xanthophylls unable their identification based only on absorption maxima, but it requires proper identification using standards and mass spectra analysis. Mass spectra of xanthophyll diesters are relatively simple and contain six types of ions above m/z 200: $[M]^+$ - the molecular ion (or

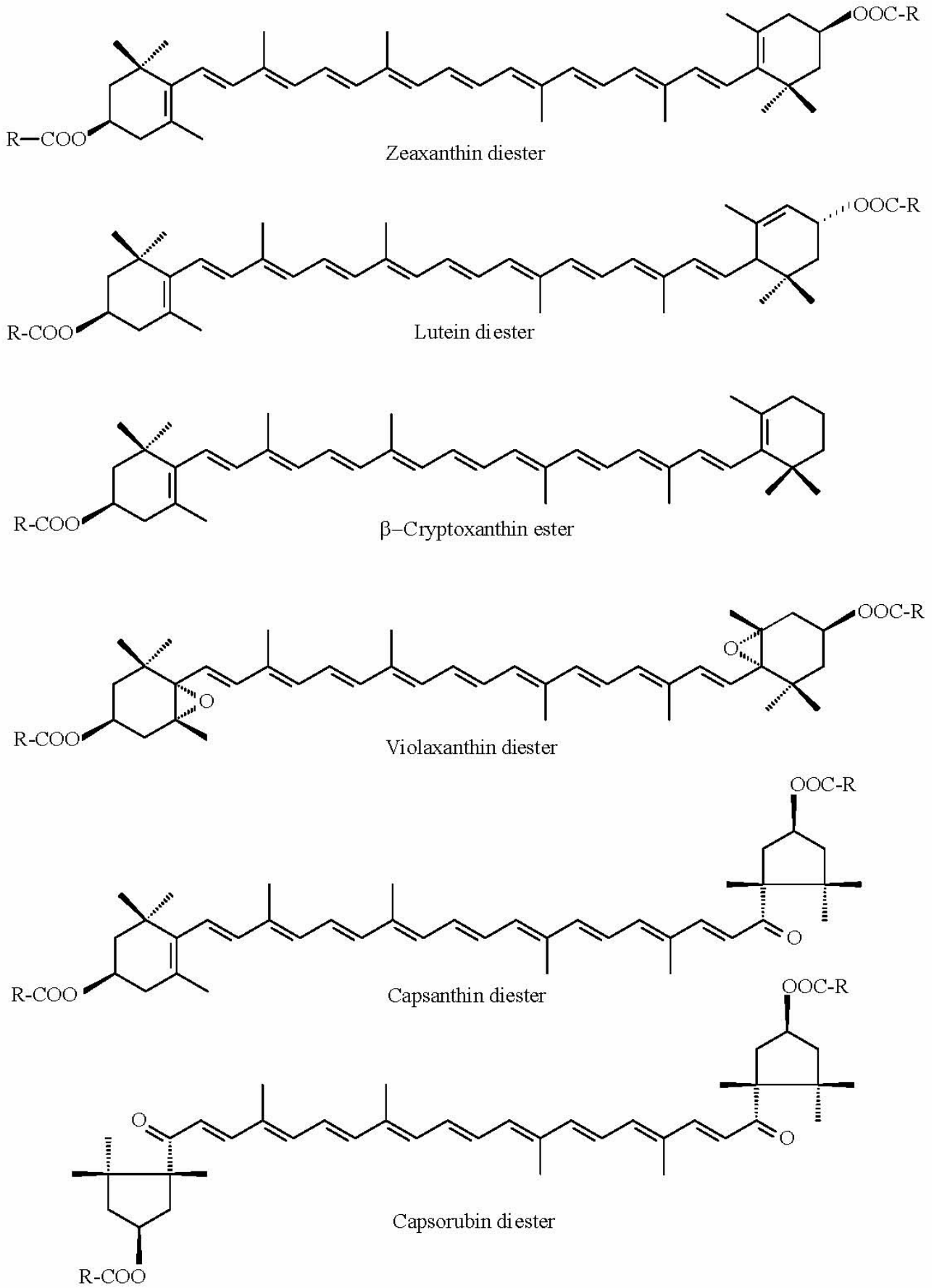


Fig. 1. Structure of major xanthophyll esters in fruits and vegetables

$[M+H]^+$ in APCI positive ionisation mode), $[M-R_1COOH]^+$, $[M-R_2COOH]^+$, $[M-R_1COOH-R_2COOH]^+$, $[R_1COOH]^+$ and $[R_2COOH]^+$. These fragments correspond to the ester molecule (ionized or protonated), to the consecutive loss of fatty acids moieties and to the fatty acids which esterify the hydroxyl group (Enzell and Back, 1995).

The esterification of xanthophyll decreases the polarity and increases the solubility in lipids. The high molecular weight and the very low polarity confer a special chromatographic behaviour to the xanthophyll esters, which is characterized by higher retention time than carotenes when separated on reversed phase stationary phase (C18 or C30). The chromatographic behaviour of esters depends on the polarity of the parent xanthophyll, on the chain length and unsaturation of the fatty acids. The retention times of carotenol esters on reversed-phase column increases as the number of carbon atoms in the fatty acids moiety increases and decreases with the unsaturation for the same chain length (e.g. linoleic acid esters of lutein elute earlier than stearic acid esters). Also, the esters of violaxanthin (which possesses epoxide group) elute in front of lutein or zeaxanthin esters (Khachik, 2009).

Xanthophyll esters can be hydrolyzed by the common alkaline hydrolysis (methanolic potassium hydroxide or sodium hydroxide) (Schiedt and Liaaen-Jensen, 1995) or by enzymatic hydrolysis with various lipases (as described in the section *Metabolism and bioavailability*) (Breithaupt et al., 2002; Breithaupt et al., 2007).

Most of the beneficial effects of carotenoids are attributed to their antioxidant activity which was proved by numerous *in vitro* and *in vivo* studies. Carotenoids are known as effective quenchers of singlet oxygen, of excited sensitizers and as scavengers of other reactive oxygen species. Antioxidant properties of carotenoids are due to the presence of polyenic system but they can be modified by the presence of different functional groups (Krinsky and Johnson, 2005; Yeum et al., 2009).

The experimental results regarding the antioxidant capacity of xanthophyll esters are contradictory. Lutein, lutein monomyristate and lutein dimyristate were found to have similar antioxidant activities when tested by methyl linoleate hydroperoxide assay and by DPPH (1,1-diphenyl-2-picrylhydrazyl) test (Subagio and Morita, 2001). In a later work, the same authors reported that lutein and its dimyristate ester have prooxidant effect on purified triacylglycerols from corn, when heated at 40°C, in darkness. However, the prooxidant activity of free lutein was more pronounced than that of its dimyristate (Subagio and Morita, 2003). Carotenoid esters of capsanthin and capsorubin from paprika exhibited a better stability toward lipoxygenase catalyzed linoleic acid oxidation than the free xanthophylls (Biacs et al., 1989). The esterified forms of capsanthin were proved to be good radical scavengers, with an antioxidant activity similar to the free form, when assessed by AMVN-induced oxidation of methyl linoleate (Matsufuji et al., 1998). Radical-induced oxidation (with AIPN as initiator) occurred at the same rate in the free and the esterified capsanthin and capsorubin, but faster for the zeaxanthin esters than for free zeaxanthin. The differences were explained by the differences in the fatty acids which esterified the xanthophylls. Capsanthin and capsorubin are

esterified with saturated fatty acids, while zeaxanthin is esterified mainly with the unsaturated linoleic acid. This proves that beside the xanthophyll, the fatty acid moiety can influence the antioxidant properties of carotenol esters (Pérez-Gálvez and Minguez-Mosquera, 2002, 2005). The scavenging capacity of hydroxyl radical, superoxide anion and singlet oxygen (determined by chemiluminescence) proved that esterification with saturated fatty acids does not affect the antioxidant activity of β -cryptoxanthin (Fu et al., 2010). Slightly lower antioxidant activity was found for the β -cryptoxanthin and zeaxanthin esters with unsaturated fatty acids (oleic and linoleic) compared to the saturated esters and to the free forms (Pintea et al., 2013a, 2013b, 2014a).

Extraction and chromatographic separation

Extraction of xanthophyll esters, as well as of all the carotenoids, requires the use of organic solvents. Several factors are taken into account when choosing the solvents, such as the structure of the matrix containing carotenoid esters and the differences in the polarities of carotenoids found in the matrix. Xanthophyll esters are non-polar compounds and usually, they are extracted with non-polar solvents. However, in different matrices the presence in the matrix of the more polar xanthophylls requires the use of mixtures also containing polar solvents. Acetone, a water-miscible organic solvent dissolves both carotenes and xanthophylls in an efficient way (Rodríguez-Amaya, 2010). Most of the studies regarding carotenoid esters extraction have included acetone as extraction solvent (Delgado-Pelayo and Hornero-Méndez, 2012; Giuffrida et al., 2013; Murillo et al., 2013). Usually, a mixture of polar and non-polar solvents or acetone alone is used for carotenoid esters extraction from different matrices. A mixture of acetone:hexane (1:1, v/v) was used to isolate β -cryptoxanthin, lutein and antheraxanthin esters from apricots (*Prunus armeniaca* L.) (Kurz et al., 2008), and for the extraction of 9-*cis*-violaxanthin and all-*trans*-violaxanthin esters from mango (Ornelas-Paz et al., 2007). Extraction of carotenoid esters from oranges juices and from sea buckthorn (*Hippophae rhamnoides* L.) was performed with a mixture of methanol:ethylacetate:petroleum ether (1:1:1, v/v/v) (Giuffrida et al., 2012; Pop et al., 2014; Weller and Breithaupt, 2003). Inbaraj et al. (2008) extracted zeaxanthin and β -cryptoxanthin esters from Goji (*Lycium barbarum*) with a mixture of hexane:ethanol:acetone:toluene (10:6:7:7, v/v/v/v). N,N-dimethyl formamide was used for xanthophyll esters extraction from potato (Fernandez-Orozco et al., 2013) and a mixture of 1,2-dichloroethane:acetone:methanol (2:1:1, v/v/v) for carotenoid esters extraction from *Capsicum annum* L. species (Daoud et al., 2014). Almost all the procedures used the addition of NaHCO₃, MgCO₃ or CaCO₃ in the first steps of the extraction in order to neutralize the acidity of plant tissues and to prevent rearrangements. The extractions are repeated until the residues are colourless and the filtrates are combined, partitioned into a separatory funnel with an adequate solvent (diethyl ether). The organic phase is removed and evaporated and the residue is further dissolved in a suitable amount of solvent for HPLC analysis (Britton, 1995).

HPLC techniques are widely used for the separation and quantification of carotenoid esters. Unsaponified extracts can be injected directly into HPLC columns or after a previous fractionation by open column chromatography (Bernhard, 1995; Pintea *et al.*, 2005; Subagio *et al.*, 1996). The first attempts to separate carotenoid esters in fruits by HPLC revealed the superiority of reversed phase (octadecylsilane, C18) over the normal phase (silica) column (Philip and Chen, 1988). Khachik and Beecher (1988a, b) developed chromatographic methods for the separation of synthetic diesters with various fatty acids of violaxanthin, auroxanthin, lutein, zeaxanthin, isozeaxanthin, and beta-cryptoxanthin on C18 reversed phase column.

Most of the separation on C18 columns used binary gradients with mobile phases containing various ratio of: acetonitrile/ethyl acetate/water; acetone/water; 2-propanol/water/ acetonitrile. Good separation of esters from sarsaparilla berries (Delgado-Pelayo and Hornero-Méndez, 2012), potatoes (Fernandez-Orozco *et al.*, 2013) and apples (Delgado-Pelayo *et al.*, 2014) were obtained on C18 RP columns. Sea buckthorn unsaponified extracts containing complex mixtures of carotenes and esters were both separated on C18 column (LiChrosorb RP 18) and on a Waters Acquity UPLC BEH Shield RP18 column (2.1x150 mm, 1.7 x150 mm, 1.7 μm) (Pop *et al.*, 2014). Octadecylsilane C18 column are robust columns which allow good separation of both unesterified and esterified xanthophylls, in a relatively short time, with affordable solvents.

C30 columns of different size are however mostly used for the separation of fatty acid esters of carotenoids in native extracts from fruits or vegetables. Binary or ternary gradients are used, based on a combination in various proportion of the following solvents: methanol/tert-butyl-methyl-ether/water (with or without ammonium acetate) or dichloromethane/methanol/acetonitrile. Good separations on C30 columns of a high number of esters were obtained for apricot (Kurz *et al.*, 2008); *Capsicum* sp. (Giuffrida *et al.*, 2013); sea buckthorn (Giuffrida *et al.*, 2012; Weller and Breithaupt, 2003); exotic fruits (Murillo *et al.*, 2013). C30 columns provide better separations, especially for long-chain hydrophobic compounds (like xanthophyll esters) and for geometric isomers (Sander *et al.*, 2000). The disadvantages of C30 columns are the longer separation time and more expensive solvents. However, for very complex matrix like sea buckthorn, pepper or citrus fruits, the method of choice is the separation on C30 column with narrow particles (3 μm).

A particular case is the analysis of carotenoids in mandarin essential oil by LCxLC, where a normal phase column (Supelcosil LC-SI) was used on the first dimension and a reversed phase (Chromolith Performance RP-18) on the second dimension. Beside other carotenoids, eighteen esters of different xanthophylls were identified (Dugo *et al.*, 2008).

Carotenoids are commonly identified based on their light absorption properties using UV-VIS and PDA detectors. Carotenoid esters have the same light absorption properties like the parent compounds and can not be identified solely based on their spectra (position of absorption maxima and the fine structure %III/II).

The availability of relatively few commercial standards led to the extensive use of hyphenated chromatographic techniques: LC/MS and LC/MALDI. Mass spectrometry detectors provide the molecular peak but also the fragments which can be further used in order to respect the minimum criteria of identification for carotenoids and their esters. Several MS detection methods were reported to use continuous flow fast atom bombardment, matrix-assisted laser desorption/ionization (MALDI), electrospray ionization (ESI), atmospheric pressure photoionization (APPI), atmospheric solids analysis probe (ASAP) and atmospheric pressure chemical ionization (APCI). APCI method can form both positively and negatively charged molecular ions species (Amorim-Carrilho *et al.*, 2014; Cacciola *et al.*, 2012; Van Breemen *et al.*, 2012; Wingerath *et al.*, 1997). The LC/APCI-MS method, characterized by high sensitivity and superior linearity of detector response, became the standard analysis method in carotenoid esters analysis. For example, 31 esters were identified in *Capsicum* sp. (Giuffrida *et al.*, 2013) and 24 esters in apple (Delgado-Pelayo *et al.*, 2014). An overview on the extraction, chromatographic methods and detections employed for characterization of unsaponified extracts, as well as the main esters identified in different fruits and vegetables are presented in Tab. 1.

Generally, the quantification of the carotenoids in carried out using the PDA detection and standard curves obtained either with commercially available standards, standards prepared by semi-synthesis (in house) (Weller and Breithaupt, 2003) or with the parent xanthophyll. Due to the difference in absorbivity, it is recommended that quantitative determination of esters is made by an internal standard calibration with isozeaxanthin esters (e.g. dinonanoate) prepared by acylation (Khachik, 2009). For the analysis of fatty acids profile in ester fraction or in matrices containing carotenoid esters, gas chromatography can be used (Pintea *et al.*, 2005; Pop *et al.*, 2013).

Biosynthesis and accumulation of carotenoid esters

Carotenoids are tetraterpenes which belong to the large family of isoprenoids. More than 32,000 isoprenoids and about 750 carotenoids have been identified so far. Most of the carotenoids are C40 isoprenoids and they are divided into carotenes (hydrocarbons) and oxygenated derivatives (xanthophylls). Carotenoid pigments are synthesized only by plants and microorganisms, while *de novo* biosynthesis of carotenoids could not be demonstrated in any animal organism (Britton *et al.*, 1998).

Similarly with other isoprenoids, carotenoids are biosynthesized by the condensation of the C5 precursors – the isopentenyl diphosphate (IPP) and dimethylallyl diphosphate (DMAPP). The biosynthesis of IPP and DMAPP takes place by two routes: the classical acetate mevalonate pathway (MVA) and the lately discovered 2C-methyl-D-erythritol-4-phosphate (MEP) pathway. The two biosynthetic pathways have different locations - the MVA pathway is localized into the cytosol while the MEP pathway is located in the plastids. In plants carotenoids are biosynthesized from C5 precursors generated mostly by MEP pathway (Moise *et al.*, 2013; Rohmer, 1998). The main steps of carotenoid esters biosynthesis are:

- head-to-tail condensation of C5 units (three IPPs and

one DMAPP) to form the C20 intermediate – geranylgeranyldiphosphate (GGPP);

- head-to-head condensation of two GGPP units, catalyzed by phytoene synthase (PSY), to form 15-cis-phytoene (a colourless carotenoid);
- desaturation and isomerization reactions to form lycopene;
- cyclization reactions, catalyzed by various cyclases, which leads to the formation of different cyclic carotene structures (β , ϵ , κ , etc.);
- chemical changes in the cycle (hydroxylation, epoxidation, etc.) to generate xanthophylls;
- esterification of hydroxyl group with corresponding acyl-coenzyme A (Britton *et al.*, 1998, Giuliano, 2014; Gomez-Garcia and Ochoa-Alejo, 2013; Moise *et al.*, 2013; Shumskaya and Wurtzel, 2013).

Significant progresses in the understanding of carotenoid biosynthesis were made during last years. Most of the enzymes involved in biosynthetic pathway were purified and characterized in several plant models like *Arabidopsis*, tomato, maize, pepper, daffodil, etc. Recently the gene encoding for the enzyme that catalyzes the xanthophyll esterification (PYP1 – pale yellow petal) was identified in tomato flowers. The PYP 1 protein has two domains, a hydrolase domain and an acyltransferase domain (LPAT – lysophospholipid acyltransferase). It was proved that the deletion or the alteration of the acyltransferase domain abolishes the esterification of xanthophylls and determined the disruption of normal chromoplast development (Ariizumi *et al.*, 2014)

Hydroxyl groups of xanthophylls are often esterified in fruits and flowers with mixtures of fatty acids. Carotenoids are biosynthesized by almost all the plastids, but xanthophyll esterification is associated with the formation of chromoplasts of fruits and flowers and with ripening in fruits (Minguez-Mosquera and Hornero-Méndez, 1994; Ohmiya, 2013). In chloroplasts, carotenoids are localized in the thylakoid membranes as component of the photosynthetic complexes. The major carotenoids in all green leaves are: β -carotene, lutein, violaxanthin, neoxanthin; while α -carotene and zeaxanthin are present in small amounts (Britton, 1995a). Xanthophylls of the chloroplasts are found in non-esterified form and the esterification occurs in senescent leaves, being associated with the degeneration of chloroplasts and the formation of chromoplasts (Biswal, 1995; Cardini, 1982).

Chromoplasts are differentiated from chloroplasts during the ripening of green fruits but also from leucoplasts or amyloplasts during ripening of other fruits or development of roots. Several types of chromoplasts have been described, with different morphologies and containing different types of carotenoid sequestering substructures, such as plastoglobules. It is considered that chromoplasts serve as a metabolic sink for the accumulation of carotenoids (Li and Yuan, 2013). Xanthophyll esters were found to be associated with fibrillins and polar lipids to form fibrils in the chromoplasts of red pepper (Deruère *et al.*, 1994). In the isolated chromoplast fibrils of red pepper all the xanthophylls were esterified in proportions varying from 75 to 100%. The authors performed an *in vitro* study to reconstitute chromoplast fibrils by mixing fibrillin with the lipid components and they found that xanthophyll

diesters (zeaxanthin and capsanthin) were the most efficient for fibril assembly. This study demonstrated the critical role of the xanthophyll ester for the fibril formation. Later, Ytterberg *et al.* (2006) identified twenty-eight proteins in the plastoglobules of red pepper chromoplasts, among them some fibrillins (structural proteins) and an esterase which might be responsible for the esterification of xanthophylls. The esterification process increases significantly the lipophilicity of xanthophylls and has an important physiological relevance for the sequestration of carotenoids in the chromoplasts (Ariizumi *et al.*, 2014; Ohmiya, 2013). It was also proved that the carotenoids stored in plastoglobules have a higher stability to light. Also, their accumulation can contribute to the protection of plastoglobules and light sensitive constituents against blue light (Merzlyak and Solovchenko, 2002).

In red pepper (*Capsicum annuum*) the xanthophylls zeaxanthin, β -cryptoxanthin, capsanthin and capsorubin are esterified with different fatty acids. Zeaxanthin, capsanthin and capsorubin, which possess two hydroxyl groups, can be present in free form, as monoesters or diesters with different fatty acids, giving rise to a complex carotenoid pattern. Zeaxanthin and β -cryptoxanthin are esterified mainly with linoleic acid (50-60 %) while capsanthin and capsorubin are esterified preferentially with middle chain saturated fatty acids. Esterification occurs not only in newly synthesized carotenoids, but also in pigment existing in green fruits, like violaxanthin (Hornero and Mosquera, 2000). Studying five red pepper cultivars, the same authors observed a 23.5 to 48 fold increase of total carotenoids during ripening process, depending on the phenotype. Also, a gradual decrease of the free (F) form of xanthophylls but an increase of partially esterified (PE, monoesters) and totally esterified (TE, diesters) fractions could be observed. Only in the last stage of ripening was a small decrease of TE noticed, as well as an increase of PE, fact that suggests the establishment of a balance between the fractions. Independent on the phenotype, the proportion of the three fractions was similar: 24.17 ± 4.06 (F), 31.48 ± 4.61 (PE) and 44.36 ± 5.05 (TE).

Similar processes were reported during ripening in apple fruits. Esters of lutein, violaxanthin and neoxanthin with various fatty acids were found in apple peel (*Malus pumila*, cv. 'Cox's Orange Pippin') (Knee, 1988). The author suggested that esters are synthesized *de novo* in ripening apples, their concentration increases during ripening independent on ethylene production so they could serve as early indicators of fruit maturity. The changes in the chlorophyll and carotenoid composition in the skin of apple fruits ('Antonovka' variety) in sun and shade ripening, both on and off the tree, were determined. Violaxanthin and neoxanthin, the major esterified xanthophylls in apples, were quantified by HPLC. In shaded apple on the three, xanthophylls esters increased from 1.4 % to 19.7 % (6 September – 11 October) while in sunlit apples there was a dramatical increase, from 5.1 % to 38.4 %. In detached fruits, xanthophylls esters reached 50 % in shaded and respectively 95 % in sunlit apples. A decrease of chlorophyll and a significant change in the carotenoid pattern was also observed, with a decrease of β -carotene and lutein and an increase of carotenoids of xanthophyll cycle (Solovchenko *et al.*, 2006). The changes in carotenoid composition of apple

skin exposed to sun light could be mediated by the reactive oxygen species generated in these conditions (Merzlyak *et al.*, 2002; Solovchenko *et al.*, 2006).

Mango fruits contain as main compounds β -carotene and dibutyrate of all-*trans*-violaxanthin and 9-*cis*-violaxanthin. The changes in the carotenoid content and the changes in the color in mesocarp and epidermis were determined for two mango cultivars 'Ataulfo' and 'Manila', after harvesting during 16 days. For both cultivars there were significant increases (exponential/second order polynomial) of all carotenoids. All *trans*-violaxanthin and 9-*cis*-violaxanthin dibutyrate reached 31.97×10^{-3} g/kg and 16.81×10^{-3} g/kg in the mesocarp of 'Manila' cultivar at the end of experiment. Interestingly, at the moment of harvest/acquisition no 9-*cis*-violaxanthin esters were detected. The increases in the concentration of all carotenoids were highly correlated with a^* and b^o color values of the mesocarp and epidermis (Ornelas-Paz *et al.*, 2008).

Sea buckthorn berries are known as rich sources of zeaxanthin and β -cryptoxanthin esters (Pintea *et al.*, 2005; Weller and Breithaupt, 2003). Depending on the region, the oily berries are normally harvested from August – September until late November. A study performed in Sweden on four sea buckthorn cultivars showed a significant increase of total carotenoids during ripening: from 323 μ g/g (28th of July), to 1015 μ g/g (15th of September). The most significant increase was recorded for the esterified fraction (from 100 to 648 μ g/g) while free xanthophylls increased only from 61 to 82 μ g/g (Andersson *et al.*, 2009). Even though sea buckthorn oil is characterized by a high content of unsaturated fatty acids, mainly oleic and palmitoleic acid (Pintea *et al.*, 2005), the carotenoid ester fraction is mainly esterified with palmitic and myristic acid. Palmitoleic acid was identified together with palmitic acid in the mixed ester of zeaxanthin, while oleic acid and other unsaturated fatty were found to esterify only lutein and in very small amount sustaining the idea of selective esterification of xanthophylls in plants (Giuffrida *et al.*, 2012). The accumulation of esters during ripening was proved by HPLC-DAD analysis in *Physalis alkekengi* L. sepals (Bunea *et al.*, 2007).

Interesting data regarding xanthophyll esters accumulation were obtained in the case of red fleshed papaya fruits (*Carica papaya* L.). Unless in the other fruits, β -cryptoxanthin esters (caprate and laurate) were already present in early ripening stages of papaya fruits and representing 42% of total carotenoids. The amount of total carotenoids increased significantly, from 130 μ g/100g in ripening stage 1 preharvest to 6214 μ g/100g in ripening stage 7 postharvest, as well as the amount of β -cryptoxanthin (free and esterified). However, the proportion of esters decreased to 19-20%, as well as the ratio β -cryptoxanthin esters/ β -cryptoxanthin. Also, the accumulation of esters seems to be dependent on the fatty acids since β -cryptoxanthin laurate accumulated with much higher rate than other esters (Schweiggert *et al.*, 2011).

Recently Fernandez-Orozco *et al.* (2014) analyzed sixty potato cultivars and found an average of 23.1% carotenoids esters. The highest xanthophyll ester content was found at 867.6 μ g/100 g dry wt, which represents 58.6% of total carotenoids. Moreover, the authors found a direct

correlation between the total carotenoid content and the ester fraction, suggesting once again the importance of esterification for the sequestration and accumulation of xanthophylls within the plastids. An interesting finding is the high proportion of linoleic acid as esterifying fatty acid (39.0%), similar with the ester fraction of tritordeum grains where lutein was extensively esterified with linoleic acid (Mellado-Ortega and Hornero-Mendez, 2012). Other fatty acids, like oleic acid (14% of total fatty acids in grains) could not be detected in carotenoid ester fraction. Based on these studies the authors suggest a high specificity of xanthophyll acyl transferase responsible for the formation of xanthophyll esters in plants. On the other hand, it was proved that the fatty acids preferentially esterify the hydroxyl group of the β -ring in the monoesters fraction of lutein (Khachik *et al.*, 1988b). A preference for esterification of lutein with lauric acid was observed in yellow varieties of raspberry (Carvalho *et al.*, 2013).

Metabolism and bioavailability/bioaccessibility

Fruits and vegetables are the major sources of dietary carotenoids for human. Due to their lipophilic character, carotenoids are metabolized following the same pathways as dietary lipids. The main steps of carotenoid metabolism are: the release from the food matrix; solubilization and incorporation into micelles; carotenoid uptake by intestinal mucosal cells and absorption (passive and active diffusion); transport in the blood by incorporation into chylomicrons and other lipoproteins; distribution and accumulation in tissues; structural modification and excretion. Bioavailability of carotenoids is affected by several factors, some of the related to the food: the food matrix and the location of carotenoids, the chemical nature of carotenoids (including isomerism), the food processing, the amount of lipid in the food, the interaction between carotenoids and with other components of the food (proteins, fibers). Additionally, the bioavailability/bioaccessibility of carotenoids is affected by human factors, like age, gender, health status (parasitic infections), habits (smokers), etc. All these factors, and probably some others (the existence of "low" or "non-responders"), contribute to a large variability in the carotenoid bioavailability in humans and in the response of human subjects to the supplementation with carotenoids (Canene-Adams and Erdman, 2008; Fernández-García *et al.*, 2012; Schweiggert, 1998).

Xanthophyll esters have to be hydrolyzed before intestinal absorption. *In vitro* studies have proved that xanthophyll esters (lutein, zeaxanthin and β -cryptoxanthin) can be hydrolyzed by pancreatic lipase, microbial lipase and by cholesterol esterase (Breithaupt *et al.*, 2002, 2007; Jacobs *et al.* 1982). In a model of simulated digestion, zeaxanthin esters purified from *Lycium chinense* were partially hydrolyzed by carboxyl ester lipase (porcine, bovine, colipase or combined) and the hydrolysis increased the efficiency of micellarization. The hydrolysis of esters also increased significantly the uptake of zeaxanthin from micelles by Caco-2 human intestinal cells (Chitchumroonchokchai and Failla, 2006).

In humans, carotenoids esters seem to be efficiently hydrolyzed before absorption. β -cryptoxanthin concentrations in chylomicrons increased after the ingestion of a tangerine concentrate rich in β -cryptoxanthin esters,

but no esters were found in blood (Wingerath *et al.*, 1995). Similarly, only small amounts of lutein esters were found in blood (Granado *et al.*, 1998) and in skin (Wingerath *et al.*, 1998) after a long term dietary supplementation with a mixture of lutein esters.

Most of the studies regarding the bioavailability of esters proved that xanthophylls esters have comparable bioavailability with the corresponding free xanthophylls and a comparable increase of plasma free xanthophylls can be observed after supplementation with food or products rich in esters. Lutein diesters were found to be with 61.6% more bioavailable than unesterified after administration of a single dose (Bowen *et al.*, 2002) but was influenced by the type of formulation (amount of oil), as reported also in a previous study (Roodenberg *et al.*, 2000). Administration of paprika oleoresin (1 g oleoresin with 5 g oil) containing esterified xanthophyll in healthy volunteers lead to a significant increase of zeaxanthin, β -cryptoxanthin and β -carotene in the chylomicrons fraction of plasma. However none of them could be identified in the esterified form, even if in the oleoresin the xanthophylls were mainly present as esters with fatty acids (Pérez-Gálvez *et al.*, 2003). Plasma β -cryptoxanthin increased significantly after administration of a single dose of both, free or esterified xanthophyll from papaya, supporting their comparable bioavailability and the effective enzymatic cleavage of esters (Breithaupt *et al.*, 2003).

A randomised, single-blind cross over study was performed on volunteers that received equal doses (5 mg in yoghurt) of zeaxanthin dipalmitate (purified from wolfberry) or unesterified zeaxanthin. The plasma samples analyzed by chiral HPLC showed a better bioavailability of esterified form, even if both forms determined a significant increase of zeaxanthin in plasma (Breithaupt *et al.*, 2004). The bioavailability of zeaxanthin dipalmitate from wolfberry was tested in a human supplementation trial in which the subjects receive 15 g/day whole fruits (containing about 3 mg zeaxanthin) for 28 days. In the supplementation group the plasma zeaxanthin concentrations increased from 0.038 to 0.096 $\mu\text{mol}/\text{l}$, demonstrating that wolfberry is a bioavailable source of carotenoids (Cheng *et al.*, 2005). The significant increase of zeaxanthin in plasma concentration after supplementation with fruits has a particular importance since zeaxanthin bioavailability from spinach or corn is low (Bone *et al.*, 2000; Mozaffarieh *et al.*, 2003). An interesting study on carotenoid bioaccessibility was performed using an *in vitro* digestion model on carrot, mango, papaya and tomato. The mentioned sources of carotenoids have a different chromoplast morphology which was supposed to influence the release of carotenoids from the food matrix. Xanthophylls and their esters accumulate in liquid-crystalline and lipid dissolved state in globular-tubular substructures of the chromoplasts in the fruits of mango and papaya. Carotenes (β -carotene and lycopene) are localized in crystalloid structures in chromoplasts of carrots, tomato and papaya. Generally, the bioaccessibility of all carotenoid was higher for mango and papaya for but xanthophylls (β -cryptoxanthin and lutein) and their esters had a superior bioaccessibility than carotenes. The addition of oil improved the bioaccessibility of carotenoids without changing the ranking. It seems that the accumulation of carotenoids (including xanthophylls

and their esters) in liquid-crystalline and lipid dissolved state of globular-tubular substructures have a positive effect on their bioaccessibility from fruits and vegetables (Schweiggert *et al.*, 2012).

Stability in model systems and food

Fruits and vegetables which provide high carotenoid intake are considered very important to maintain health and to reduce the risks of several degenerative diseases or cancer (Krinsky and Johnson, 2005; Rao and Rao, 2007). The carotenoid content in plant derived food is influenced by several factors, such as genetic origin, environmental conditions, cultivation practice and post-harvest treatment. Post-harvest treatment, including transport, storage, processing and cooking, is a key step in preserving the nutritional value of plant food. Carotenoids are more stable in vegetables and fruit than in isolated (purified) form. In plant tissues, carotenoids are protected by molecular interaction with other compounds (proteins, lipids) and by the presence of other antioxidants. Disruption of the tissues exposes carotenoids to light, oxygen and oxidizing enzymes like lipoxygenase (Britton and Khachik, 2009; Caris-Veyrat, 2008; Mercadante, 2008). Xanthophyll esters, as all the carotenoids, are susceptible to degradation under high temperature, light, acidic pH and reactive oxygen species. The mechanisms involved in carotenoids degradation in food are similar with the degradation of pure compounds and occur mainly through oxidation, geometrical isomerisation, but also by rearrangement of 5,6-epoxides and dehydration. Enzymic oxidation is due to the contact of carotenoids with lipoxygenase which leads to the bleaching of pigments. Non-enzymic oxidation occurs as a consequence of direct interactions of carotenoids with dioxygen molecule or with reactive oxygen species (superoxide anion radical, hydroxyl radical, singlet oxygen and hydrogen peroxide). The non-enzymic oxidation of carotenoids results in the formation of different epoxides and apocarotenals. High temperature, light and low pH determines the geometrical isomerisation of carotenoids. The position of the double bond of the polyene chain which is isomerized and also the proportion between the *Z* isomers which are formed depends on the applied treatment (Britton and Khachik, 2009; Caris-Veyrat, 2008; Mercadante, 2008).

The stability of free carotenoids in model systems or in different food systems were reviewed in very comprehensive publications (Maiani *et al.*, 2009; Mercadante, 2008). Compared to free carotenoids, the studies on the stability of xanthophylls esters are relatively scarce and the approaches used are very different. Thermal stability of xanthophyll esters was evaluated using purified esters or on food containing esters. Using purified lutein mono- and dimyristate Subagio *et al.* (1999) found that they were more stable toward heat treatment and UV light than free lutein, suggesting that esterification with saturated fatty acid stabilize the highly unsaturated lutein. More recently, the thermal stability of β -cryptoxanthin esters with saturated fatty acids compared to free β -cryptoxanthin was determined in tetrahydrofuran solutions during incubation at 50 °C, in dark, for up to 72 hours. The stability of pigments, determined by HPLC-PDA, showed that β -cryptoxanthin esters were more stable than free β -

cryptoxanthin (Fu *et al.*, 2010). Thermal stability tests were performed at different temperatures on β -cryptoxanthin and zeaxanthin esterified with saturated (myristic and palmitic acids) and unsaturated fatty acids (oleic and linoleic acids) compared with the free xanthophylls. At all temperatures, the esters with saturated fatty acids showed a significantly better stability than the free xanthophylls, while the esters with unsaturated fatty acids exhibited lower stability than the saturated ones, but still higher than the free β -cryptoxanthin and zeaxanthin (Bunea *et al.*, 2013; Pintea *et al.*, 2013). Lutein esters (mono- and dimyristate) were found to be more stable toward UV light than free lutein (Subagio *et al.*, (1999). Contrarily, the photostability toward UVA irradiation was found to be lower for β -cryptoxanthin palmitate than for free cryptoxanthin, both in hexane solution and in liposomes containing carotenoids. The degradation rate was higher for carotenoids in liposomes and the highest degradation rate was recorded for β -cryptoxanthin palmitate (Arita *et al.*, 2004).

Relatively few papers were published on the stability of xanthophyll esters in fruits or vegetables. Khachik and Beecher (1988a, b) found that xanthophylls esters in squash were more stable than free carotenoids. Diesters of violaxanthin were more stable than monoesters during cooking and no significant isomerization nor epoxide rearrangement were observed. Stability studies were performed on red peppers (*Capsicum annuum* L.) and hot chilli peppers (*Capsicum frutescens* L.) pods minced, heated for 5 min at 80, 90 or 100 °C and then lyophilized. Capsanthin, capsorubin, zeaxanthin and β -cryptoxanthin mono- and diesters from both types of peppers showed comparable photo- and thermal stability, in all the cases higher than for non-esterified forms (Schweiggert *et al.*, 2007).

A different approach was applied for investigation of thermal lability of carotenoids of tamarillo fruits (*Solanum betaceum* Cav). In this study, the samples were heated (at 80, 90, 95 °C) in hermetically sealed amber vials, in both degassed and not degassed nectar, in which the final level of oxygen was determined. HPLC-PDA/MS analysis of samples before and after treatment showed that zeaxanthin esters appeared to be the less thermo-labile carotenoids, in both experimental conditions (Mertz *et al.*, 2010).

Most of the stability studies performed on pure esters (isolated or synthesized), in esters fractions or in plant materials rich in xanthophyll esters demonstrated the protective effect of esterification, especially with saturated fatty acids.

Conclusions

Xanthophyll esters represent an important fraction of total carotenoids in many fruits and vegetables. The major xanthophylls present in esterified form are lutein, zeaxanthin, β -cryptoxanthin, violaxanthin, capsanthin and capsorubin. Commonly consumed fruits and vegetables, like apples, apricots, mandarins, mangoes, papayas, red and chilli peppers, potatoes or squash, contain xanthophylls mainly in esterified form. Some other rich sources have been identified, like: wolfberry (goji), sea buckthorn, persimmon and exotic fruits (sapote, maracuja, frutita, mamey, sastra, corozo, tamarillo, etc).

Identification of carotenoid esters in very complex unsaponified extracts requires powerful chromatographic techniques and mass spectrometry detectors. Significant progresses were reported during last years due to the technical development of separation and identification tools. Also, there is an increasing interest in the characterization of unsaponified extracts which, in contrast to the saponified ones, reflects the native composition of food. Lutein esters have been identified as major compounds in squash; zeaxanthin esters are present in wolfberry, sea buckthorn and red pepper; β -cryptoxanthin esters in apricots, sea buckthorn, mandarin; violaxanthin and neoxanthin esters in apple and capsanthin and capsorubin esters in pepper.

Esterification of xanthophylls has physiological significance for the chromoplast formation. The increased lipophilicity of esters is important for the sequestration of carotenoids and for the formation of specialized structures (plastoglobules, fibrils) in the chromoplasts. The esters also provide protection against photooxidative damages in plants. Esterification process and accumulation of esters occurs during ripening in fruits and is associated with significant change in the colour of fruits. Even if the specific enzymes which catalyze the esterification process were not characterized yet, detailed analytical data regarding the carotenoid composition suggests a selectivity of these enzymes for certain fatty acids and selectivity for the ring in the case of non-symmetric carotenoids like lutein.

Xanthophyll esters seem to be efficiently hydrolyzed and absorbed in humans leading to a comparable bioavailability to the unesterified compounds. Also, the xanthophyll esters preserve the antioxidant capacity of the parent compounds while having a better stability in fruits during storage and processing. All these properties are important from the perspective of the use of fruits rich in xanthophyll esters as valuable components of the human diet and as sources of bioactive compounds in the prevention of severe degenerative diseases.

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